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Conservation of mass &
proportions, counting
atoms related terms.

Terms in this set (28)

law of conservation of
mass. states that mass is
neither created nor

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destroyed during
ordinary reactions or
physical changes.

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Name CHAPTER 3

TEST continued Date

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BLANK Write the

correct term (or terms)

in the space provided. 9,

If a particular compound

is composed of elements

A and B, the ratio of the

mass of B to the mass of

A will always be the

same. This is a

statement of the law of

exactly 12 g of

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Carbon-12 is referred to
as a(n) 11.

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Atoms: The Building
Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the
following questions in
the space provided. 1.

Explain the difference
between the mass

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number and the atomic number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

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Answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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~~Answers~~

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CH 3 Protons-Neutrons-

Electrons-Mass

#-Atomic # CH 3

Atomic #, Mass #, P-N-

E Chart II. CH 3

Review Atomic Theory

. CH 3 Test

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conservation of mass.

mass is neither created

nor destroyed during

ordinary chemical

reactions or physical

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changes $2H_2 + O_2 = 2H_2O$

O. law of definite proportions. Holt

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Atoms: The Building Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the

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following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

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Chapter Test Name

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A, continued Use this

figure to answer

questions 7 and 8.

_____ 7. A solution containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C .

According to information in the figure, this temperature change would result in

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a. an additional 5 g of Li
2SO₄ in ...

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MULTIPLE CHOICE I.

In the modern periodic table, elements are arranged according to

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Chapter Review. p.59.

Problem Bank. YES!

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