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L10-10L 10.591 mL of Titrant
Used: 10.09L/9.95L10.00L mmole
of NaOH: 1.009 moll 9.95 no 1.0
ml mmole of H₂SO₄ 0.5 mal 5.0
0.5 5.0nd Molarity of H₂SO₄
Solution 0.05 M 0.5 M 0.05 M
0.05M Average Molarity 0.6M.

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Grades: The lab component counts as 25% of your final grade in CHEM 103. Per department policy, you must perform every experiment in order to pass CHEM

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Reaction 1: $1 \text{ NaHCO}_3 (\text{s}) \rightarrow 1$

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NaOH (s) + 1 CO₂ (g) Reaction 2:
2 NaHCO₃ (s) 1 Na₂CO₃ (s) + 1
CO₂ (g) + 1 H₂O (g) Reaction 3:
2 NaHCO₃ (s) 1 Na₂O (s) + 2.
Propose a voltaic cell that would
coat a piece of copper with
platinum. Include the half
reactions for your cell and the

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voltage potential based on the.

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treatment facility, which operates

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Answers - Maharashtra

1. Identify the group for each unknown. Your unknowns will be from three different groups.

(Example: A K^+ salt, a halogen,

and an acid.) 2. Identify the specific compound from each group. (Example: KCl , Br_2 and

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HCl) 3. Explain the reasoning that enabled you to come to these conclusions.

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important operations in a chemistry laboratory is the massing of objects. Since chemistry is an exact science, the massing of substances which enter or result from a chemical change must be done with the best possible ...

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